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Synthesis of β -peroxy-lactones using 30% $H_2O_2^{\Rightarrow}$

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Abstract—Cyclopropyl carbinols 1a—e react with H_2O_2 in the presence of concd H_2SO_4 at 0–5 °C to furnish the corresponding hydroperoxides 2a—e in 45–82% yields. This efficient trapping of cyclopropylmethyl cations by the hydroperoxy group has been exploited to prepare compounds 14–23, a new series of cyclopropyl substituted β-peroxy-lactones. © 2005 Elsevier Ltd. All rights reserved.

The cyclopropylmethyl cation and its rearrangement have been extensively studied in the context of cyclopropane chemistry. Cyclopropylmethyl substrates solvolyze with abnormally high rates forming the cyclopropylmethyl cation as a key intermediate, that either rearranges to the corresponding cyclobutyl cation or a homoallylic cation. Thus the products of solvolysis often include compounds not only derived from unrearranged cyclopropylmethyl cations but also from cyclobutyl and homoallylic cations. On the basis of solvolysis experiments the order of stability of these cations follows the order: cyclopropylmethyl cation > cyclobutyl cation > homoallylic cation.

The greater stability of the cyclopropylmethyl cation is due to the presence of the cyclopropyl group. The bonds in cyclopropane cannot be pure σ bonds and have considerable sp^2 character and therefore provide a π cloud, which stabilizes the positive charge on the adjacent carbon atom. The rearrangement of the cyclopropylmethyl cation to a homoallylic cation has been exploited synthetically to convert cyclopropyl carbinols into homoallylic alcohols, homoallylic acetates and homoallylic halides. In the course of work designed to prepare homoallylic hydroperoxides we discovered that cyclopropyl substituted carbinols react efficiently with 30% H_2O_2 in the presence of concd H_2SO_4 to furnish cyclopropyl substituted hydroperoxides in good yields.

Thus cyclopropyl carbinols 1a–e, easily accessible from cyclopropyl methyl ketone by reaction with an appropriate Grignard reagent, reacted with 30% H₂O₂ in the presence of concd H₂SO₄ at 0–5 °C to furnish hydroperoxides 2a–e in 45–82% yields. A similar reaction of 1a with t-BuOOH furnished the corresponding tert-butyl peroxide 2f in 57% yield (Scheme 1). 5.6 Thus, H₂O₂ acts as an efficient trap for the cyclopropylmethyl cation before it rearranges to cyclobutyl or homoallylic cations.

We have extended this observation to the preparation of β -peroxy-lactones. Thus, β -hydroxy esters **4–13** prepared by reaction of cyclopropyl ketones **3** with α -halo esters/zinc were treated with 30% H_2O_2 in the presence of concd H_2SO_4 in THF at 0–5 °C to furnish β -peroxy-lactones **14–23** in 30–80% yields (Scheme 2, Table 1). Peroxy-lactones **15** and **21** were isolated as an inseparable mixture of diastereomers. 8

a, R= Ph; **b**, R=4-ClC₆H₄; **c**, R=3-MeOC₆H₄;

d, R=n-pentyl; **e**, R=cyclohexyl

Scheme 1.

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Scheme 2. Reagents and conditions: (a) α -bromo ester, Zn, benzene, reflux, 2–7 h; (b) 30% H₂O₂, concd H₂SO₄, THF, 0–5 °C, 4–6 h.

Table 1. Yields of β-peroxy-lactones 14–23

Compound	\mathbb{R}^1	\mathbb{R}^2	\mathbb{R}^3	Yield (%)
14	Me	Н	Н	80
15	Me	H	Me	38
16	Cyclopropyl	Н	H	69
17	Cyclopropyl	H	Me	39
18	Cyclopropyl	H	Et	35
19	Phenyl	Н	H	42
20	4-Chlorophenyl	H	H	55
21	Me	Н	Et	52
22	Me	Me	Me	33
23	Cyclopropyl	Me	Me	30

β-Peroxy-lactones have earlier been prepared from β-hydroxy acids using 90% H_2O_2 . Since the preparation of 90% H_2O_2 is hazardous, the present method is advantageous since commercially available 30% H_2O_2 is used. However, the presence of the cyclopropyl group is essential for this reaction as β-hydroxy esters **24**, **25** and **26** failed to react with 30% H_2O_2 under the same conditions (Scheme 3).

Since there was no report on the antimalarial activity of β -peroxy-lactones we also tested these easily accessible β -peroxy-lactones against *P. falciparum* (NF-54) in vitro¹⁰ (Table 2).

Compounds 14 and 15 which showed activity at 10 µg/mL were also tested in vivo¹¹ against multi-drug resistant *P. yoelii* in mice by the intramuscular (i.m.) route (Table 3). Both these compounds showed significant suppression of parasitaemia. However, considering the high order of activity shown by artemisinin and its derivatives, the activity shown by these compounds is weak.

While various classes of peroxides such as 1,2,4-triox-anes, ¹² 1,2,4,5-tetraoxanes, ¹³ endoperoxides, ¹⁴ H₂O₂, ¹⁵ alkyl hydroperoxides, ¹⁶ peroxyamines, ¹⁷ peroxyketals, ¹⁸

Scheme 3. Reagents and conditions: (a) 30% H₂O₂, concd H₂SO₄, THF, 0-5 °C, 7-9 h.

Table 2. In vitro antimalarial activity of β-peroxy-lactones **14–23** against *P. falciparum* (NF-54)

Compound	$MIC^{a,b}$ (µg/mL)	
14	10.00	
15	10.00	
16	>50.00	
17	50.00	
18	50.00	
19	50.00	
20	>50.00	
21	>50.00	
22	>50.00	
23	>50.00	
Artemisinin	0.03	
Chloroquine	0.04	

^a MIC = Minimum concentration inhibiting development of ring stage parasites into the schizonts.

Table 3. In vivo antimalarial activity of compounds **14** and **15** against *P. yoelii* in Swiss mice by the i.m. route

Compound	Dose (mg/kg/day)	% Suppression on day 4 ^a	No. of surviving mice on day 28
14	192	76.60	0/5
15	192	Toxic	0/5
	96	77.52	0/5

^a Percent suppression = $[(C - T)/C] \times 100$; where C = parasitaemia in the control group, and T = parasitaemia in the treated group.

etc. have been tested for antimalarial activity, to our knowledge this is the first report on antimalarial activity of β -peroxy-lactones.

In conclusion, we have discovered that hydrogen peroxide is an efficient trap for cyclopropylmethyl cations and have exploited this observation to develop a convenient route for the preparation of β -cyclopropyl substituted β -peroxy-lactones using 30% H_2O_2 . Work related to trapping of cyclopropylmethyl cations by other soft nucleophiles such as thiols is currently in progress and will be reported in our subsequent publications.

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 $[^]b$ 50.00 µg/mL was the highest concentration used in this study.

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- In sharp contrast to the reaction reported here, spirocyclopropylcarbinols 27 have been reported to react with 90% H₂O₂ under acid catalysis to give exclusively bicyclic hemiacetal peroxides 28. [Lillie, T. S.; Ronald, R. C. *J. Org. Chem.* 1985, 50, 5084–5088.]

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- 7. Typical procedure for the synthesis of β-peroxy-lactones: To an ice-cooled (0–5 °C) solution of 4 (0.5 g, 2.90 mmol) in THF (15 mL) was added 30% H₂O₂ (2 mL) followed by dropwise addition of concd H₂SO₄ (2 mL) with constant stirring and then at 0–5 °C for 5 h. The reaction mixture was diluted with cold water (50 mL) and the aq layer extracted with ether (3 × 50 mL). The combined organic layers were washed with saturated aq NaHCO₃ solution (50 mL) and water (2 × 25 mL). The combined organic layers were dried over anhyd Na₂SO₄ and the solvent was evaporated under vacuum to give the crude product, which on column chromatography over silica gel using benzene-hexane (1:1) as cluant furnished 14 (0.33 g, 80%).
- benzene-hexane (1:1) as eluant furnished 14 (0.33 g, 80%). 8. Selected spectral data: Compound **2a**: oil; IR (neat, cm⁻¹): 3399; ¹H NMR (200 MHz, CDCl₃) δ 0.35–0.40 (m, 2H), 0.50-0.58 (m, 2H), 1.28-1.41 (m, 1H), 1.45 (s, 3H), 7.24-7.51 (m, 5H); 13 C NMR (50 MHz, CDCl₃) δ 1.60, 2.79, 19.09, 21.99, 87.19, 126.55, 127.82, 128.75, 143.87; FABMS (*m*/*z*) 179 (M+H)⁺. Compound **2f**: oil; IR (neat, cm⁻¹): 762, 878, 1365, 1449; ¹H NMR (200 MHz, CDCl₃) δ 0.31–0.51 (m, 4H), 1.23 (s, 9H), 1.27–1.35 (m, 1H), 1.45 (s, 3H), 7.21–7.50 (m, 5H); 13 C NMR (50 MHz, CDCl₃) δ 0.00, 0.88, 18.57, 20.99, 25.34, 77.33, 81.78, 125.00, 125.21, 126.13, 143.63; FABMS (m/z) 235 $(M+H)^+$. Peroxylactone 14: oil; IR (neat, cm⁻¹) 1796; ¹H NMR (200 MHz, CDCl₃) δ 0.47–0.64 (m, 4H), 1.02–1.20 (m, 1H), 1.43 (s, 3H), 2.85 (s, 2H); ¹³C NMR (50 MHz, CDCl₃) δ 0.00, 0.43, 16.05, 20.91, 41.19, 87.32, 173.89; EIMS (m/z) 142 (M^+) . Peroxy-lactone 15 (mixture of diastereomers): oil; IR (neat, cm⁻¹) 1791; ¹H NMR (200 MHz, CDCl₃) δ 0.49–0.63 (m, 4H), 0.95–1.11 (m, 1H), 1.19 and 1.35 ($2 \times s$, 3H), 1.22 and 1.34 ($2 \times d$, 3H. J = 7.4 Hz each), 3.01 and 3.05 (2 × q, 1H, J = 7.4 Hzeach); ¹³C NMR (50 MHz, CDCl₃) δ 0.00, 0.19, 0.35, 0.99, 8.52, 9.03, 12.94, 15.88, 16.21, 20.77, 46.23, 47.14, 89.40, 90.08, 177.07, 177.35; FABMS (m/z) 157 $(M+H)^+$. Peroxylactone **16**: oil; IR (neat, cm⁻¹) 1798; ¹H NMR (200 MHz,

- CDCl₃) δ 0.55–0.61 (m, 8H), 1.02–1.13 (m, 2H), 2.78 (s, 2H); ¹³C NMR (50 MHz, CDCl₃) δ 0.00, 0.23, 15.32, 39.20, 88.77, 173.95; FABMS (m/z) 169 (M+H)⁺. Peroxylactone **21** (mixture of diastereomers): oil; IR (neat, cm⁻¹) 1789; ¹H NMR (200 MHz, CDCl₃) δ 0.50–0.65 (m, 4H), 0.87–1.04 (m, 1H), 1.09–1.20 (m, 6H), 1.50–2.04 (m, 2H), 2.79–2.88 (m, 1H); ¹³C NMR (50 MHz, CDCl₃) δ 1.41, 1.66, 2.27, 2.36, 12.14, 12.22, 14.24, 16.72, 17.98, 19.36, 19.59, 22.79, 54.27, 54.58, 90.57, 91.27, 177.94, 178.22; FABMS (m/z) 171 (M+H)⁺.
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- 10. The compounds 14–23 were evaluated against *P. falcipa-rum* (NF-54) using a minor modification of the technique of Rieckmann et al. ¹⁹ The asynchronous parasites obtained from cultures of *P. falciparum* were synchronized after 5% sorbitol treatment so as to contain only ring stage parasites. ²⁰ A parasite suspension in medium RPMI 1640 at 1–2% parasitaemia and 3% hematocrit was dispensed into the wells of sterile 96-well plates. Test compounds were serially diluted in duplicate wells to obtain the final test concentrations. The culture plates were incubated in a candle jar at 37 °C for 36–40 h. Thin blood smears from each well prepared at the end of incubation period were microscopically examined and the concentration, which inhibited the maturation of rings into the schizont stage, was recorded as the MIC.
- 11. The in vivo efficacy of compounds 14 and 15 was evaluated against *Plasmodium yoelii* (MDR) in the Swiss mice model. The colony bred Swiss mice (25 \pm 1 g) were inoculated with 1×10^6 parasitized RBC on day zero and treatment was administered to a group of five mice at each dose, from days 0 to 3, in two divided doses daily. The drug dilutions were prepared in groundnut oil, so as to contain the required amount of the drug in 0.1 mL and administered intramuscularly for each dose. Parasitaemia levels were recorded from thin blood smears between days 4-28. 21
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